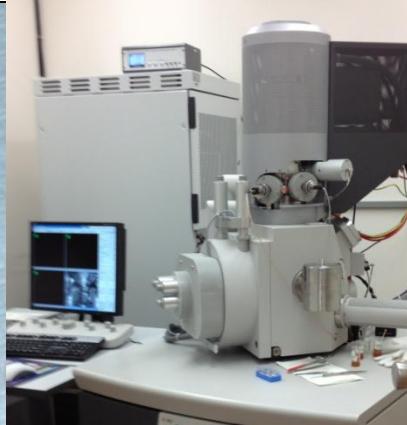


# Mobility, Availability and Reaction Products of P from MAP, DAP and APP Fertilizers



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and  
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# Introduction

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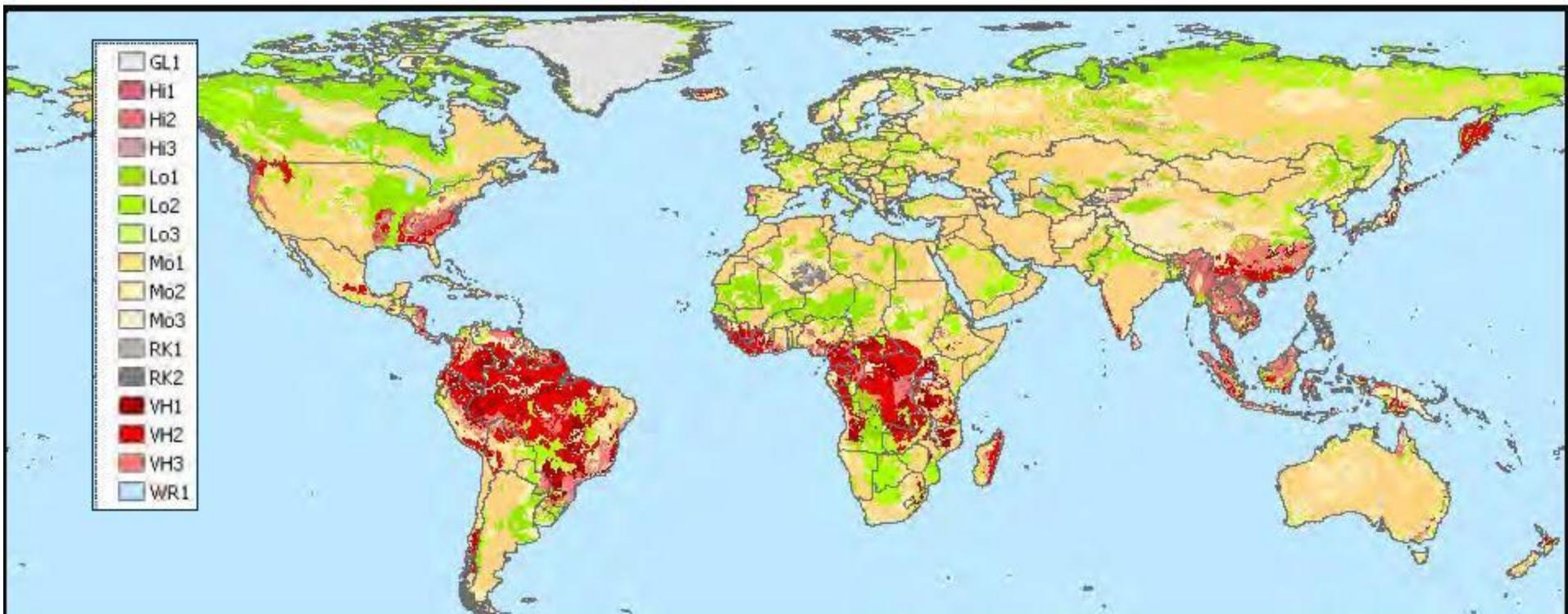
- A high proportion of applied P is rapidly converted to insoluble phosphates that plants have virtually no access
- These transformations mainly depend on
  - fertilizer sources
  - type of soil
  - soil moisture conditions
- Understanding major reaction products of fertilizer P in different soil types and their solubility may help designing better suited fertilizer formulations for different soil types

# Objectives

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- To understand mobility, availability and reactions products of P from MAP, DAP and APP fertilizers in different soils
- To integrate mobility, wet chemical data and reaction products to understand their performance in different soils

# How widespread or distributed are soil containing high P fixing capacity?



International Soil Reference and Information Centre, 2011

- ❖ There is a great need in increasing P availability on these enhanced P fixing soils.

# Soils

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## Six soils

1. Oxisol, Brazil
2. Calcareous soil, Idaho
3. Andisol, Ecuador
4. Calcareous soil, South Australia
5. Ultisol, Sri Lanka
6. Calcareous or acid soil, KS

# Methodology

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## Treatments:

- ❖ Control
- ❖ MAP (11-52-0 N-P<sub>2</sub>O<sub>5</sub>-K<sub>2</sub>O)
- ❖ DAP (18-46-0 N-P<sub>2</sub>O<sub>5</sub>-K<sub>2</sub>O)
- ❖ APP (11-37-0 N-P<sub>2</sub>O<sub>5</sub>-K<sub>2</sub>O)
- ❖ MAP + Fertilizer Enhancement Product (FEP)
- ❖ DAP + FEP
- ❖ APP + FEP



Petri dish size: 87 mm

MAP: 42 mg granules per dish; DAP: 43.4 mg granule; APP: ~40  $\mu$ L  
N was balanced by Urea

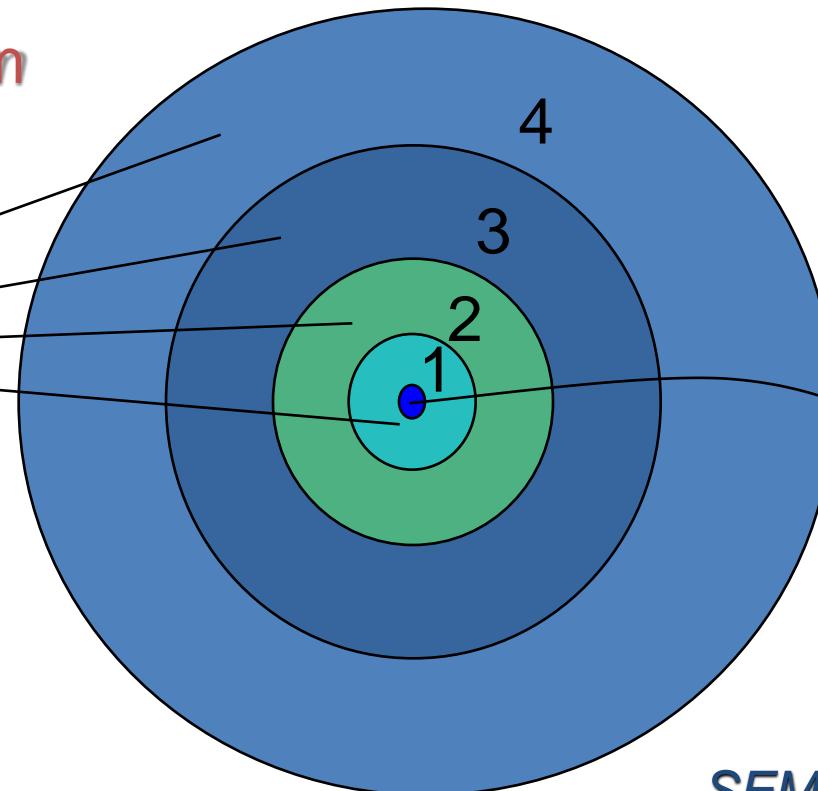
# Methodology (cont.)

## Soil Sampling and Analyses

Diffusion of P from  
fertilizers

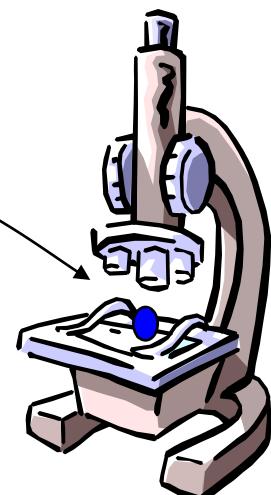
Samples

- pH
- Total P
- Resin P
- Reaction products (XANES)



43.5 mm  
25  
13.75  
7.5

Reaction products  
in granules

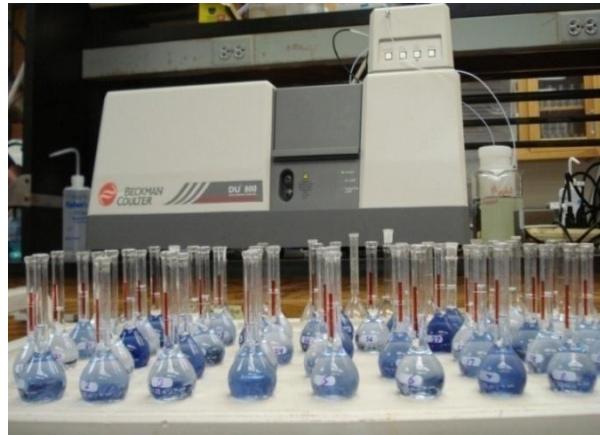


- SEM-EDX and/or XRD
- Changes in structure
  - In situ elemental analyses

# Methodology (cont.)

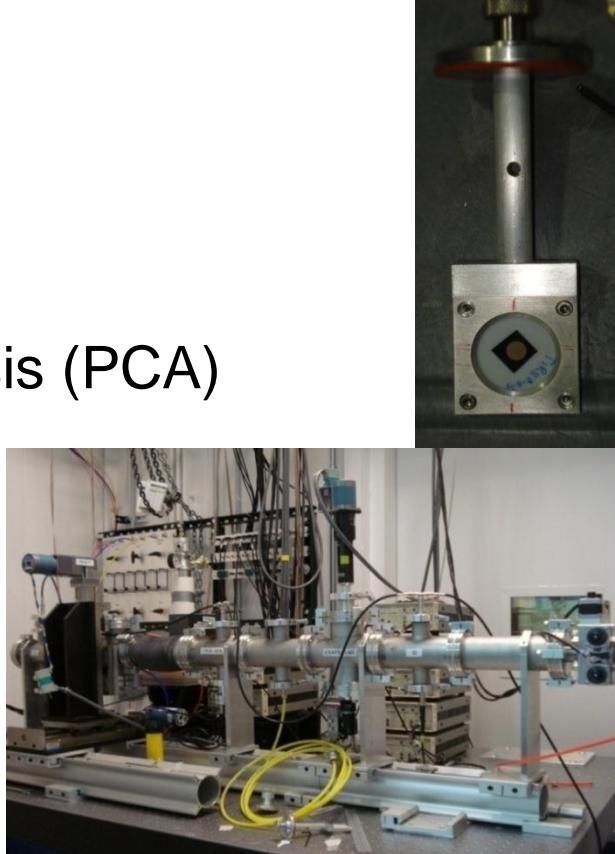
## Wet chemical based analyses

- Soil pH (1:5 soil:water)
- Total P –Aqua-regia digestion
- Resin extractable P



## P Speciation and granule observations

- X-ray Absorption Near Edge Structure Spectroscopy (XANES)  
Data analysis: Principal Component Analysis (PCA)  
followed by Linear combination fitting-LCF
- Scanning Electron Microscopy- Energy Dispersive X-ray Analysis



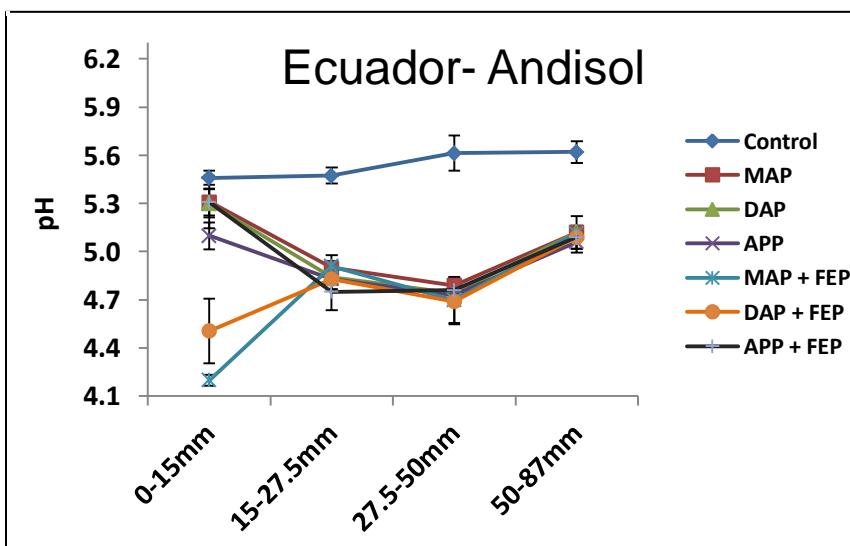
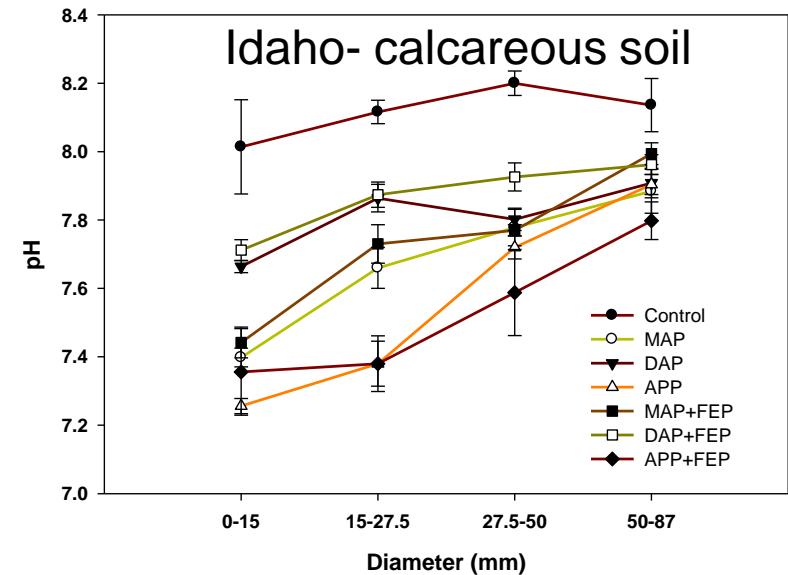
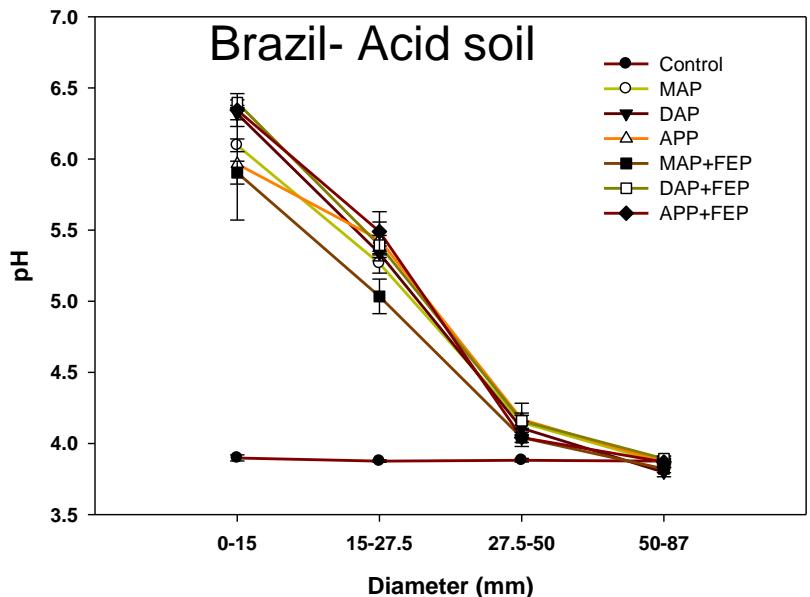
# Selected Soil Properties

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Sample ID	pH (H <sub>2</sub> O)	CEC cmol <sub>c</sub> /kg	OM %	Ext. Ca	Ext. Fe	Ext. Al	Total P
----- mg/kg -----							
Brazil soil	4.3	12.4	3.7	49	52.9	79.5	237
Idaho soil*	8.0	19.6	0.6	3376	2.4	n.d.	468
Ecuador soil	5.5	16.5	5.4	565	37.6	6.5	447

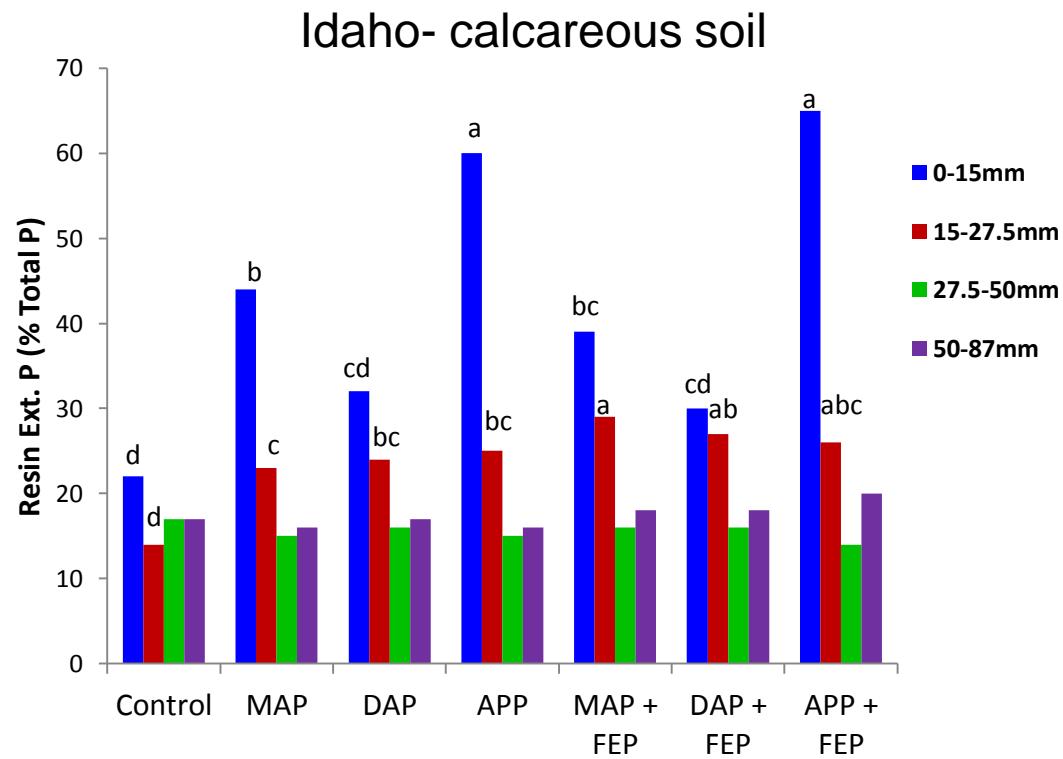
\* CaCO<sub>3</sub> = 7.4%

# Soil pH



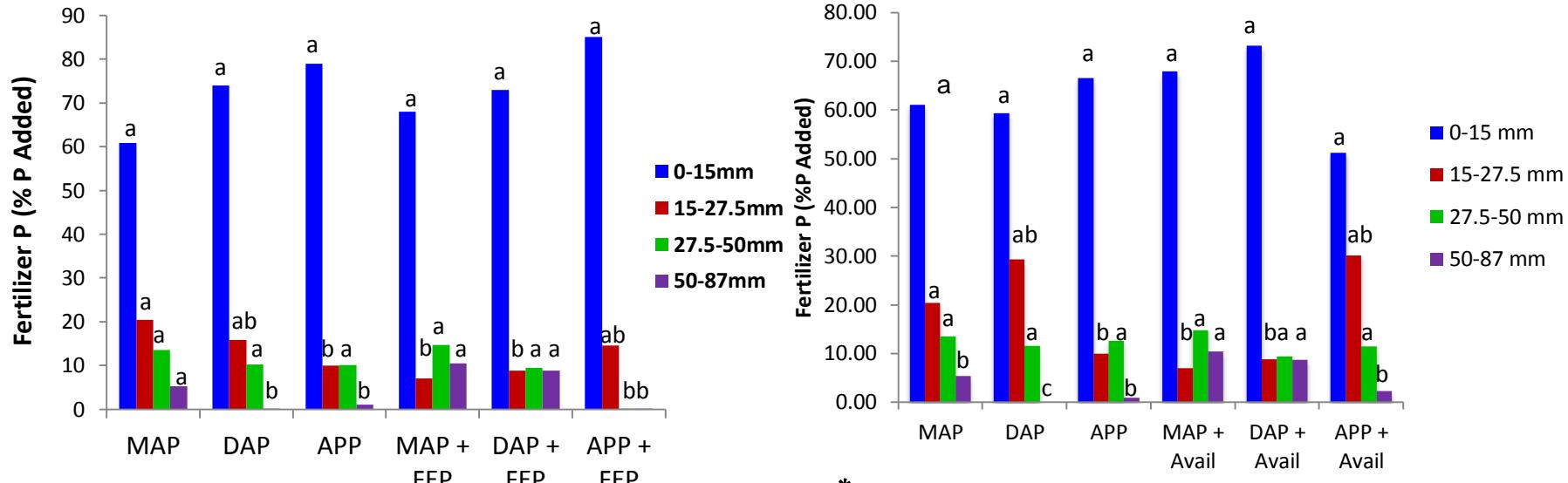
# Resin P

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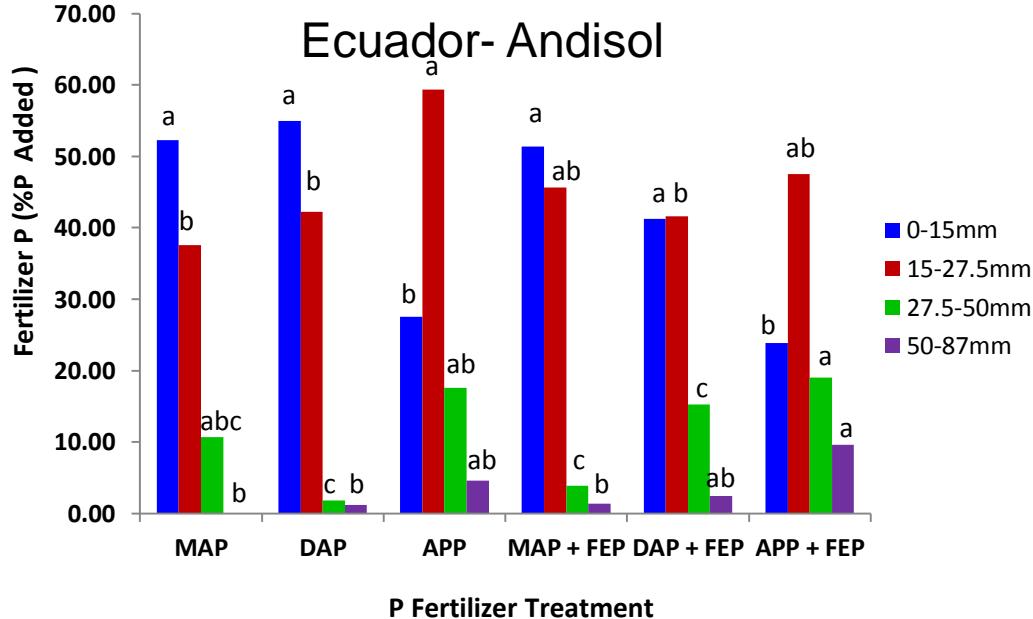


# Fertilizer P Distribution

Brazil- Acid soil      Idaho- calcareous soil

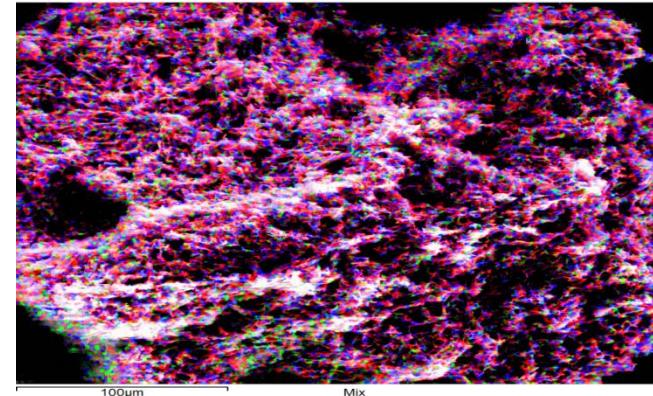
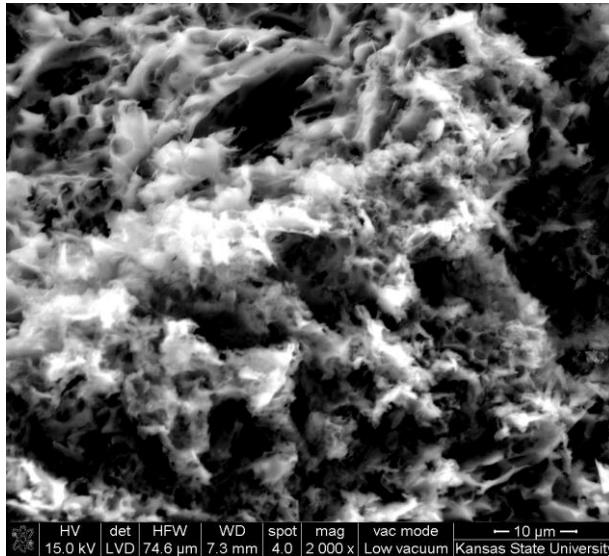
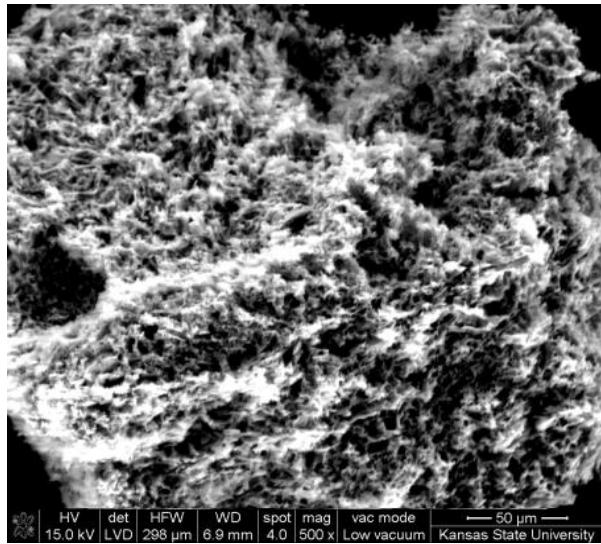


\*

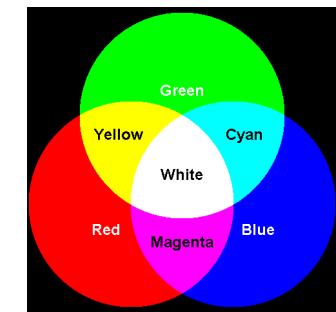


# MAP Granule- Incubated 3 days

## Ecuador Soil

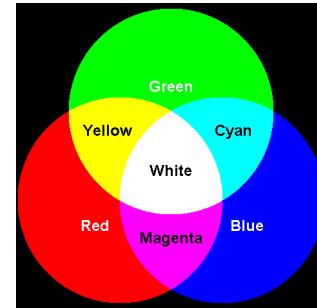
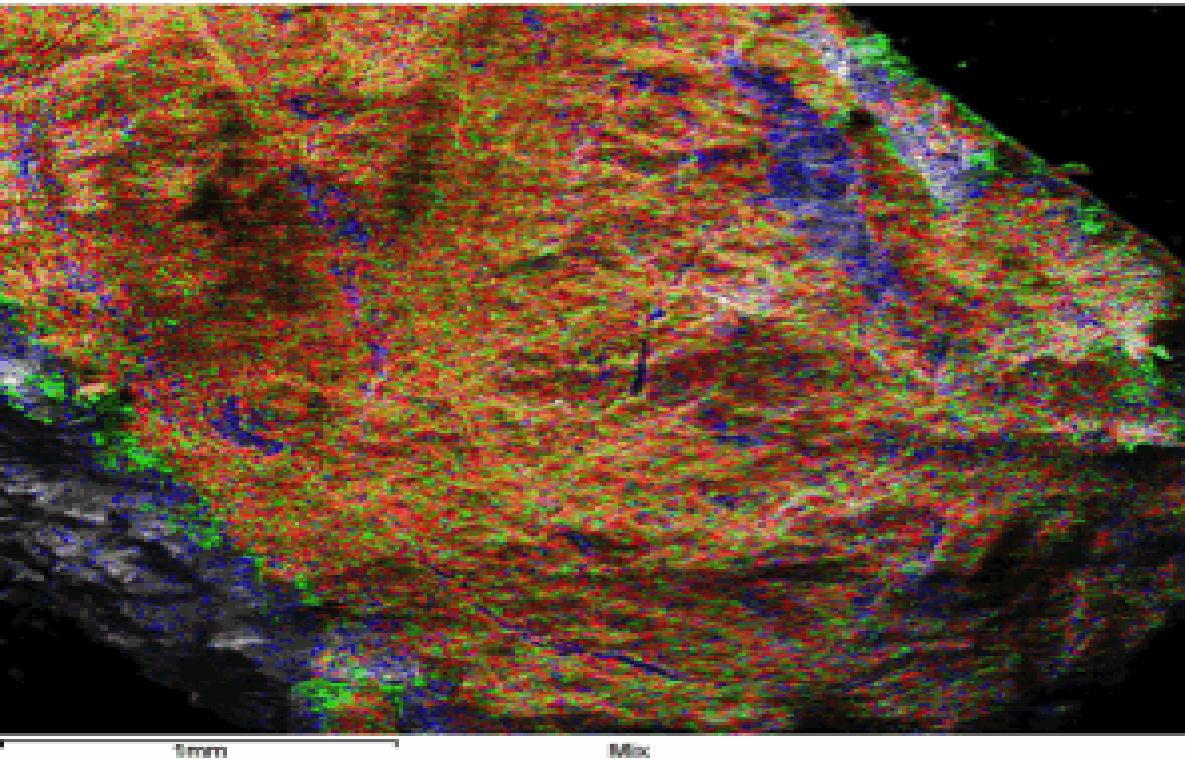


Elemental map-EDS  
P=Red  
Al=blue  
Fe=Green



Element	Wt %
N	-----
Al	4.07
P	12.51
Fe	3.86

# MAP Granule- Incubated 5 wk Ecuador Soil

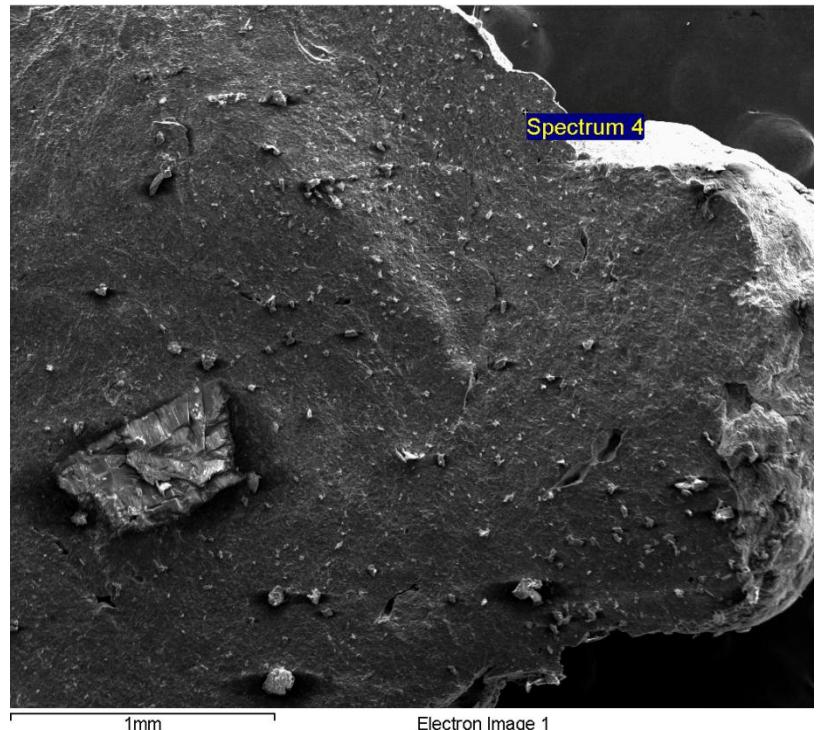
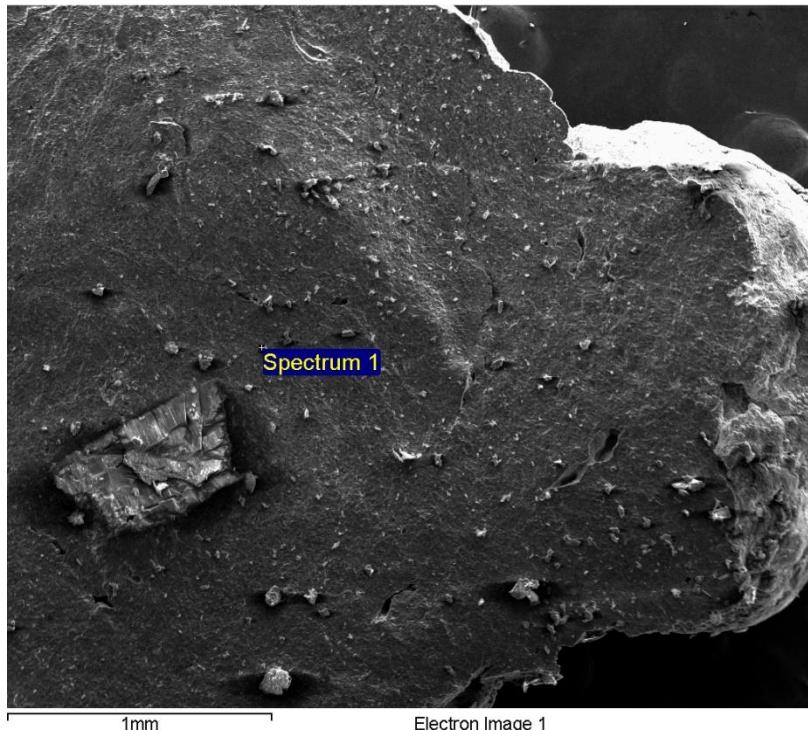


N	-----
Al	4.55
P	11.81
Fe	4.54

Zn, Re, Mg  
~ 1- 1.5 %

# DAP Granule- Original

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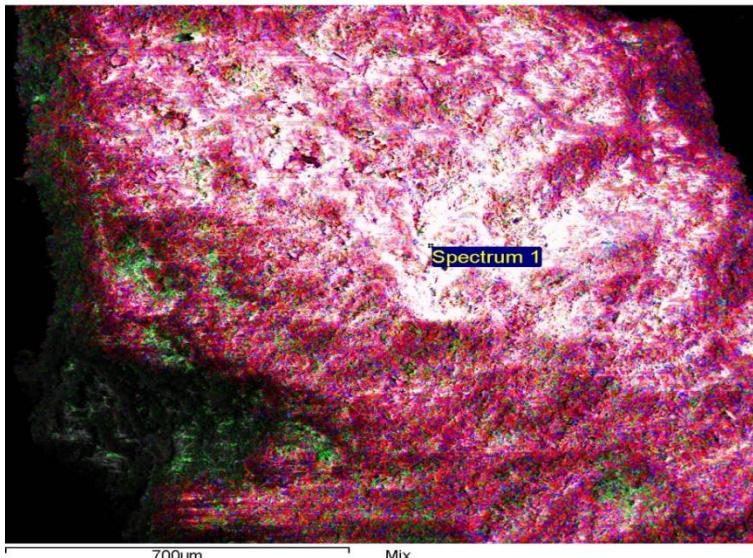
Element/Wt %

N	14.9
Al	0.8
P	21.7
Fe	0.5
Ca	0.2

Element/Wt %

N	13.3
Al	0.6
P	19.2
Fe	0.3
Ca	0.1

# DAP Granule- Incubated 5 wk Ecuador Soil



Elemental map-EDS

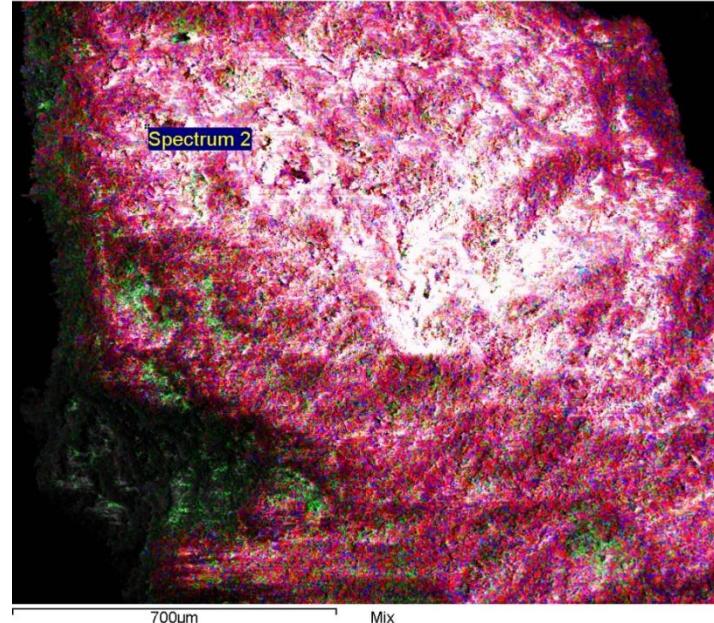
P=Red

Al=Blue

Fe=Green

Element-Wt %

N	-----
Al	5.5
P	12.9
Fe	4.9
Ca	5.3



Elemental map-EDS

P=Red

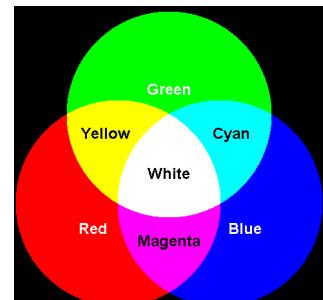
Al=Blue

Fe=Green

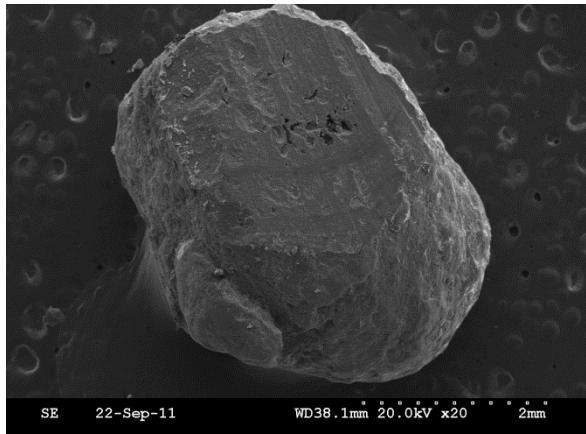
Zn, Re, Mg  
~ 1%

Element-Wt %

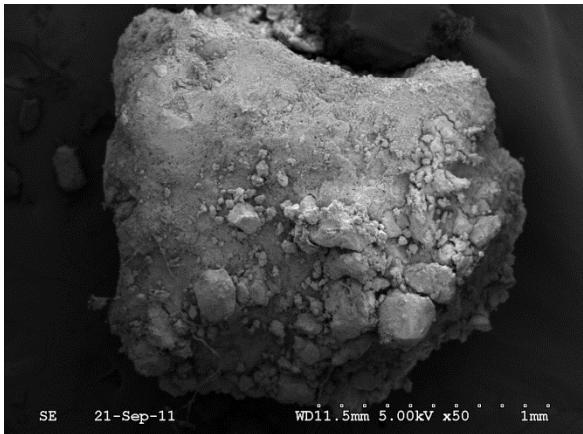
N	-----
Al	5.8
P	13.3
Fe	4.3
Ca	4.9



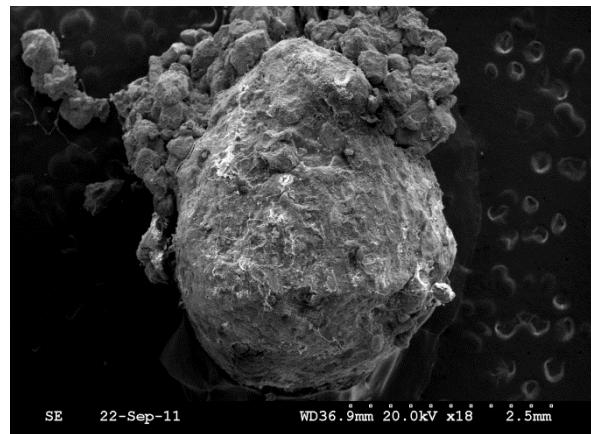
# SEM Images of Granules



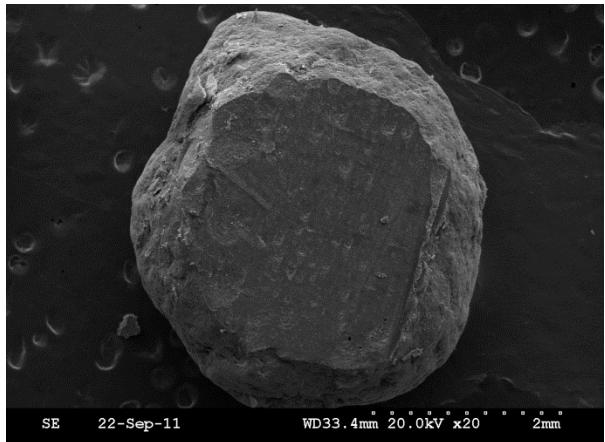
Original MAP granule



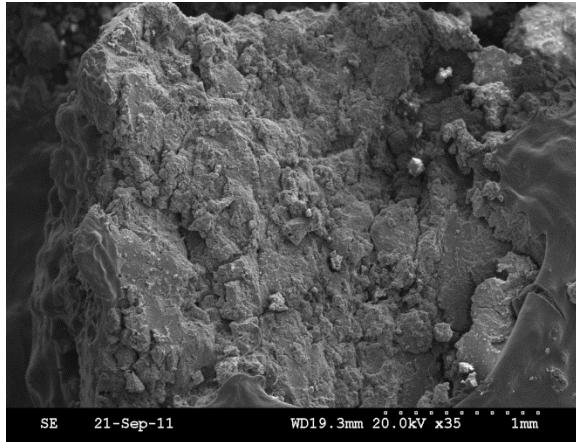
Incubated MAP granule- Brazil soil



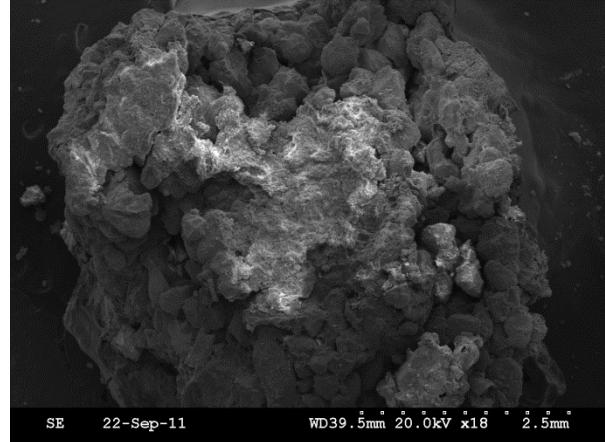
Incubated MAP granule- Idaho soil



Original DAP granule



Incubated DAP granule- Brazil soil



Incubated DAP granule- Idaho soil

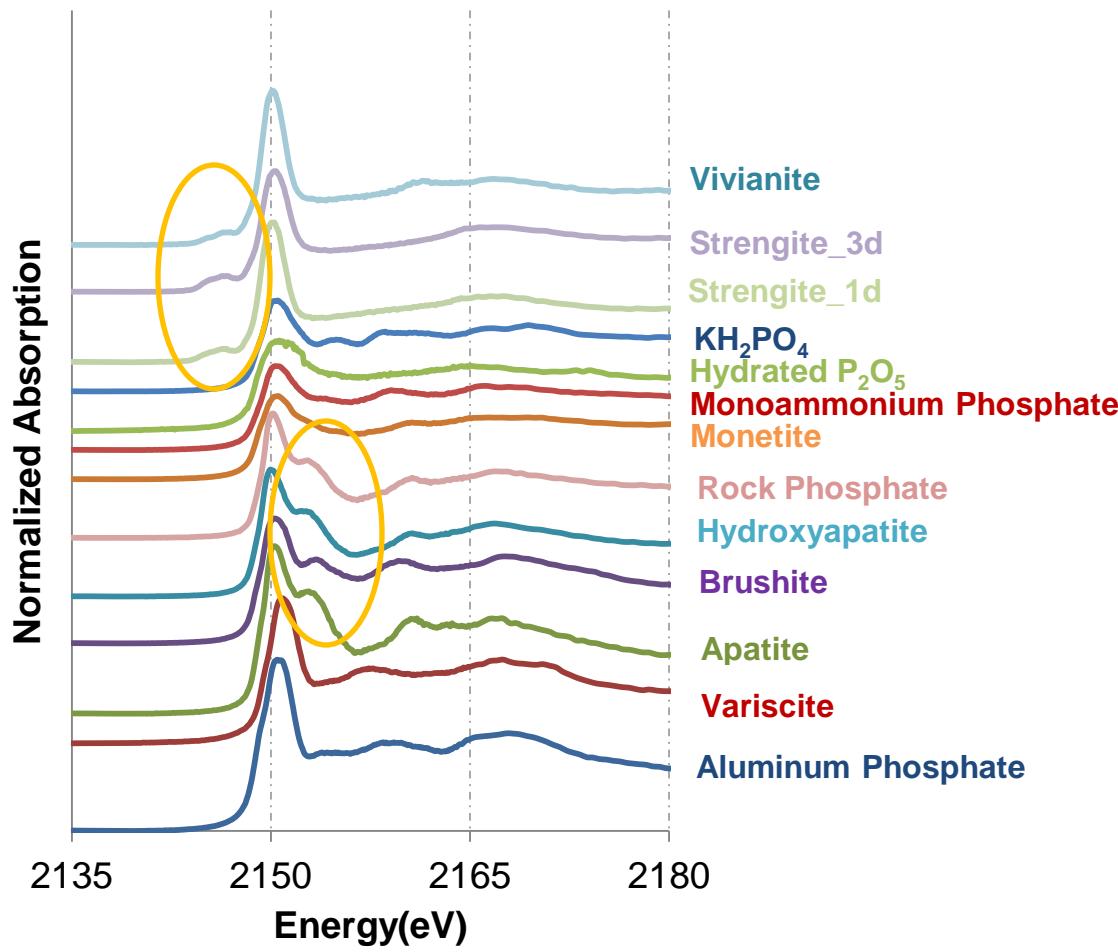
## Brazil Acid Soil

	N	Mg	P	Ca	Al	Fe
----- % by weight -----						
DAP Control	14.0	0.5	26.6	0.4	0.9	1.0
DAP Incubated	3.0	1.6	12.6	1.8	6.4	11.2
MAP Control	10.7	0.8	31.8	1.1	1.1	0
MAP Incubated	0	2.3	15.3	0.8	6.4	6.0

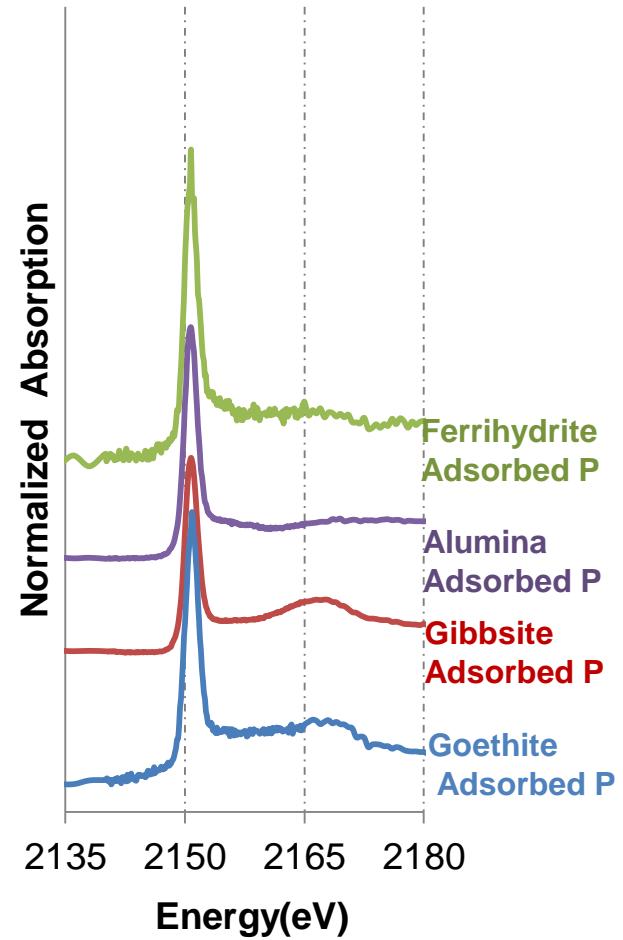
## Idaho Calcareous Soil

	N	Mg	P	Ca	Al	Fe
----- % by weight -----						
DAP Control	14.0	0.5	26.6	0.4	0.9	1.0
DAP Incubated	0	1.4	13.8	13.5	5.3	4.9
MAP Control	10.7	0.8	31.8	1.1	1.1	0
MAP Incubated	0	1.3	11.5	10.2	4.1	5.5

# X-ray absorption near-edge structure (XANES) spectroscopy analysis



Normalized P K-XANES spectra of standards used for LCF fitting



# XANES Data- Brazil acid soil

## P species in soil (%)

Section I- 0 to 7.5 mm

Treatment	Aluminum Phosphate	Alumina Adsorbed P	Ferrihydrite Adsorbed P	Iron Phosphate	Red. Chi Square
Control	13.9	-	64.1	21.9	0.27
MAP	-	-	72.1	27.9	0.32
DAP	-	47.3	-	52.7	0.04
APP	-	43.6	-	56.4	0.02
MAP + FEP	-		24.1	75.9	0.02
DAP + FEP	-	33.7	-	66.3	0.01
APP + FEP	-	21.4	-	78.6	0.00

Is Al-P less available compared to Fe-P in acid soils?

# XANES Data- Idaho Calcareous soil

## P species in soils (%)

Section I- 0 to 7.5 mm

Treatment	Calcium Phosphate	Aluminum Phosphate	Ferrihydrite Adsorbed P	Vivianite	Red. Chi Square
Control	68.6	-	31.4	-	0.06
MAP	59.2	-	31.0	9.8	0.02
DAP	64.1	-	35.9	-	0.04
APP	27.8	-	48.2	24.0	0.04
MAP + FEP	43.2	-	36.6	20.2	0.06
DAP + FEP	57.1	-	42.9	-	0.06
APP + FEP	32.0	7.1	61.0	-	0.16

APP and APP+FEP: Less Ca-P and/or more ferrihydrite sorbed P

# Summary

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- Most of the fertilizer P remained within the first two rings of the Petri dishes
- Enhanced P diffusion was observed for both APP and APP-FEP treatments
- Soil pH measurements at 5 wk alone do not provide explanations for enhanced P diffusion in some soils
- Osmotic effects of fluid fertilizer may not be detrimental to outward movement of P- thereby reducing chances of P precipitation

# Summary (cont.)

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- In calcareous soil, a possible explanation for increased resin extractability for the APP, APP-FEP treatments- increase in acidification immediately around fertilizer application point which resulted in decrease in calcium-P species

# Acknowledgements

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- Kansas State University- P Fellowship Consortium of International Plant Nutrition Institute; Mosaic, Agrium, Potash Corp., and JR Simplot, and Fluid Fertilizer Foundation for funding
- E. Francisco, L. Prochnow and T. Tindall for providing soils
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